TEST IS COMING

- Test on Friday, 9/22
 - Ch.1 through Ch.5



- Multiple choice, bring #2 pencil, eraser, & your ACT testing skills
- No need for calculator
- No phones
- Work through the practice test posted on the class webpage, end of chapter stuff in Mastering

ORBITS AND ENERGY

Total orbital energy = gravitational potential energy + kinetic energy



ESCAPE VELOCITY



Fig.4.15

PLAY

- Add more kinetic energy than there is gravitational potential energy, and things don't orbit: they escape!
- These "unbound" orbits are parabolas or hyperbolas rather than ellipses

ORBITS AND THE CENTER OF MASS



- If the Earth pulls on the Moon – by the 3rd law, the Moon should also pull on the Earth
 - Earth goes around Moon too!
- A *binary* system like the earth and moon spins about the *center of mass* of the system

ANOTHER PREDICTION FROM NEWTON:

- If all bodies attract one another, shouldn't the planets attract each other too?
 - Newton predicts yes, but much less than from the much heavier Sun
 - Jupiter should slow down Saturn as Saturn passes
- Newton asks Royal Astronomer John Flamsteed if this happens
 - It had been observed!
 - ... and later, this effect was used to discover Neptune

CHANGING ORBITS

- So how do you change an orbit?
 - Add or subtract kinetic energy from the thing in orbit
- Add:
 - burn rocket fuel
 - Steal energy from some other object with gravity
- Subtract:
 - Friction from atmosphere
 - Give energy to some other object with gravity
 - Or burn rockets pointed backwards to slow down

EXAMPLE: BOOSTING TO A NEW

Works
 both
 ways:
 can fire
 rocket
 to slow
 down

ORBIT

escape velocity

Rocket starts in low orbit.

Firing the rocket long enough gives it enough extra orbital energy to escape Earth.

> Firing the rocket a little gives it extra orbital energy, raising it to a higher, more elliptical orbit.

Fig.4.18

CASSINI, TODAY!

- This morning at 6:54am the Cassini spacecraft did a final rocket burn to slow down, and crashed into Saturn
- In Ch.8: all the great Saturn pictures came from this spacecraft over the past 13 years



Follow this link for a very cool 3D Cassini Grand Tour webpage from National Geographic

EXAMPLE: GRAVITATIONAL INTERACTION

Works both

 ways: come in
 ahead of or
 behind the big
 thing?

Fig.4.18



OTHER EFFECTS



PLAY

Tides

- The Moon's pull on the Earth's Ocean is *differential*
 - Stronger near to Moon
 - Weaker far from Moon
- Oceans bulge out
- Earth spins under bulge
- We see Ocean rise and fall twice per day

MORE TIDES



- The pull of the Sun also makes tides
 - A smaller effect than the moon, but...
 - if pull of Moon and Sun line up, extra large spring tides happen
 - if Moon and Sun are pulling sideways, extra small *neap tides* happen
 - The Earth itself stretches tidally too!
 - At most about 9 inches

MORE MOON EFFECTS

- All that sloshing about creates tidal friction
 - Slows down rotation of Earth (25 ns/day)
 - Lets moon slip away (1 cm/year)
- On Moon, tide-like action has *tidally locked* one side of the moon (the denser side)
 - The same side of the Moon always points at the Earth
- Precession
 - Moon pulls slightly squashed Earth over, causes axis to precess once each 26,000 yrs (like a top)



SAME SIDE OF THE MOON

- We always see the same side of the moon
 - why?



View back at Earth and the far side of the moon as seen by NASA's DSCOVR spacecraft

THE MOON IS OFF-BALANCE

 Like a Weeble or Punching Bag, the heavier side points "down"



Moon and ISS from Austrailia by Dylan O'Donnell



LUNAR LIBRATIONS

- Over time, tidal friction slows the wobbles until we barely notice them
 - Still there, we call them "Librations"
 - Lets us see 59% of the moon's surface

"Lunar libration with phase Oct 2007" by Tomruen from wikipedia

LIGHT AND MATTER

How can we tell what's going on up there?

MODELS AS TOOLS

- So far, we've watched the theory of gravity and planetary motion get assembled
- In Ch.5, we will see many models presented in their current form
 - Use them as tools
 - They also were carefully put together
 - But we'll not follow their story completely
- All have been tested carefully, make accurate predictions, and have convincing theories of "why" backing them up

WAVE MOTION



- Waves in water pass by a stick
- The water doesn't travel – the wave does
- Stick bobs up and down

WAVE ANATOMY

- Wavelength length of wave
- Amplitude height of wave
- Period
 - how long for one complete wave to pass you
- Frequency
 - How many complete waves pass you per second?





ELECTRIC FIELDS

- Opposite charges attract, like charges repel
- The force which does this can be thought of as lines
- Wiggle a charge the line wiggles





A wave

- Wiggling Electric and Magnetic fields
- Has wavelength λ
- Frequency f
- Travels at $c = \lambda x f$
- c constant at 186,000 mi/sec (300,000 km/s)
- Electromagnetic Radiation
 - Visible light is a small part of this
 - Comes in tiny bits called "photons"

SPEED, WAVELENGTH, AND FREQUENCY



WAVELENGTH, FREQUENCY, AND ENERGY

 $\lambda \times f = c$ $\lambda = \text{wavelength}, \quad f = \text{frequency}$ $c = 3.00 \times 10^8 \text{ m/s} = \text{speed of light}$

> $E = h \times f =$ photon energy $h = 6.626 \times 10^{-34}$ joule × s

THE HIGHER THE PHOTON ENERGY...

- a. the longer its wavelength.
- b. the shorter its wavelength.
 - Energy is independent of С. wavelength.



a.

IMPORTANCE FOR ASTRONOMY

- We can't go out there to take samples of stuff or run experiments
- Light contains all the information we will ever get from most of the universe
- So, we must understand light, and how it can be used to find out about where it was made and what happened to it on the way here

MATTER



10⁻¹⁰ meter

Ten million atoms could fit end to end across this dot.

Atom: Electrons are "smeared out" in a cloud around the nucleus.

Nucleus: Contains positively charged protons (red) and neutral neutrons (gray).

Fig.5.5

ATOMIC TERMINOLOGY

atomic number = number of protons
atomic mass number = number of protons + neutrons
(A neutral atom has the same number of electrons as protons.)







atomic number = 1 atomic mass number = 1 (1 electron)



- atomic number = 2 atomic mass number = 4
 - (2 electrons)



Carbon (¹²C)

atomic number = 6 atomic mass number = 12 (6 electrons)

MORE TERMINOLOGY

- Isotope: same # of protons but different # of neutrons (⁴He, ³He)
- Molecules: consist of two or more atoms (H₂O, CO₂) stuck together



atomic number = *number of protons*